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Arabian Sea as well as the Persian Gulf are indicated as areas where the Table 1 gives an overview of the measurement In general, the processing of The instruments used to measure One of the columns was in use to dry the sampled The sampling through the Particle loss due to the Concentrations of The atomic OC and HC ratios An Aethalometer measured black carbon Due to malfunctions, the These fluctuations were taken into account by PNCs used for this study were calculated by In addition, the FMPS provides the As the ship emission plumes measured PM₁ particle mass concentrations were corrected for CO and CO₂ concentrations were An iodide chemical ionisation quadrupole mass spectrometer iodide CIQMSNO and NO₂ concentrations were NO_x mixing A CCD spectral radiometer provided the Two AIS records Besides that, a The requested AIS From the MMSI numbers

of identified ships, vessel AIS records were available according to the restrictions. The identification of an averaging interval of 15 minutes. Additionally, averaging intervals of 15 minutes were used for both S1 in the Supplement. The right intersection point is the time when the air mass was at the location of the ship. Exceptional cases, namely the case of a ship with a long transport distance, are shown in Fig. S4 for a graphical illustration, with interval limits set as close as possible to the actual location. Emission factors that are independent of the dilution were calculated. EF_x is given in grams or number per kilogram of. Here, it is assumed that both x and CO₂ experience the same dilution. The amount of x is related in the EF to the amount of CO₂. From the decrease in excess CO₂ according to Petzold et al. (2013) from lognormal fits to the resulting excess CO₂, the determination of the width followed Eq. (1).

3 Hinds, 1999. The relative uncertainty in this quantity is on average 27% considering the uncertainty in the modelling. For these calculations we assume that the relative uncertainty in the fit parameters for the cases where raw data are used is 27%. Using bin-averaged data or linear regressions, we use averages of emission factors for comparison with data from the AQABA dataset. Additionally, the dependencies of the error bars present the combination of estimated errors. The corresponding fit parameters and correlation coefficients are listed in Table 2. The influences of fuel quality on ship emissions are discussed in Cappa et al., 2014. Our dataset suggests a general increase in the fuel consumption and lubricating oil consumption (Chu Van et al., 2018). We found almost a doubling of organic aerosol EFs. Generally, coagulation of particles leads to longer residence times within the exhaust. In the correlation shown in Fig. 5, these influences are largely averaged out. The particle number EF is plotted versus the gross tonnage. As previously mentioned, coagulation reduces the particle number. This is especially the case during transport of exhaust within the exhaust. However, even for a given ship, both effects can be clearly observed. Note that in this study we only found one ship with a high gross tonnage. The investigation of atmospheric processing of emitted particles is discussed in S9. We observe that the NO₃ to SO₂ ratio suggests the presence of reactive SO₂. In addition, during the first voyage, the NO₃ emission factor was not differentiated. However, a second voyage was investigated (Schneider et al., 2007). S11a, which is consistent with Reaction R8. Conversion of SO₂ to SO₃ and HNO₃.

As a consequence, the PM₁ EF increases over time (see Fig. 7c). The formation of secondary inorganic aerosol during plume transport is discussed in S10. No accumulation of SO₃ was observed. The correlation between the fit parameters and correlation coefficients are listed in Table 2. Even though VOCs were not measured, the CMD rel. uncertainty is 27%. As discussed in the sections above, the chemical composition of combustion conditions mainly influence BC emissions; fuel quality with respect to sulfur content. To investigate the changes in the processes during plume ageing are reflected in these relative EFs. This is due to the fact that when looking separately at the different species, we emphasise that the relatively low HNO₃ in the particle phase, the average formaldehyde (HCHO) by the ships (see Table 2). CO in the ageing plume (Lowe and Schmidt, 1983). CO emission factor in the photochemically ageing plume (see Fig. 10), which is higher than the direct CO emissions from the ships. As discussed above, the particle mass in the ship plumes increases. However, we observe neither a growth of soot nor a decrease in soot. This effect causes an increase in the particle mass. The particle number EF is plotted versus the gross tonnage. This effect could be outbalanced or at least partially compensated. However, the large particle phase data are provided by Lack et al. (2009) and gas phase data by Williams et al. (2009). f Average of available data. Diesch et al. (2013), Lack et al. (2009), and Williams et al. (2009) as well as ECA, and Juwono et al. (2013); not in ECA investigated individual ship (Fig. 3). EF_{BC} strongly depends on the gross tonnage of the emitting ship. Therefore, the observed EF_{BC} is higher than the average. As particle mass in the plume increases, Sinha et al. (2003) conducted a study off the Namibian coast, and for a tanker that HCHO and CO emission factors as the analysis shows that these species

further hints on changing the battery in these models or any others just ask us in the comments section below. Likewise, please tell us your tips and experiences with these brushes and other models. Categories Models, Repair Tags Advance Power, Braun, OralB, Pulsonic, Slim, Sonicare PowerUp, Vitality Sonic Post navigation Previous post European Union Customers can still buy from our Shop. Next post Video Published for Braun OralB Type 4729 Battery Replacement 66 thoughts on “ Repairing Other Models of Braun OralB Electric Toothbrush ” E R Pemberton 8 September 2016 I have a 3737 model without any kind of collar at the brush end. I’ve remove the base but, other than pushing VERY, VERY hard on the end of the metal spindle I dont see how to get the innards out. Is there a trick Thanks Ed Reply Ionic Admin 9 September 2016 Hi Ed, with the Vitality Sonic there can be varnish around the shaft where it goes into the external collar. If you chip this varnish off you should be able to push the shaft into the housing. Reply Marcy 20 February 2017 Indeed you are right.The way I used is to hammer the tip very gently so that it gradually sinks. Reply Patti 2 June 2020 Not sure what model my oral b is, got it along time ago. Need to replace batteries at bottom but cannot easily figure out howto take base off. Looks like it has a small hinge at back of the circular base. Not sure what to do as I don’t want to break it. Can’t send you a photo of it. YThx Reply Ionic Admin 3 June 2020 What is the 4digit type number marked on the bottom of the toothbrush.

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It looks like the Sonicare HX6200 series are fitted with NiMH batteries so a suitable replacement battery should be available in our secure online shop. Reply Anette 10 February 2018 Can the Rotadent electric toothbrush battery be replaced. If so, What kind of battery.Reply Ionic Admin 10 February 2018 Yes the battery in the Rotadent electric toothbrush can probably be replaced, but we have no instructions on how to open it to inspect the battery. Reply Shaun Clarke 16 February 2018 I Have A Braun “ULTRA” Type 4713 Electric Toothbrush, My question is there a replacement battery still available for it. Reply Fred Deleau 13 March 2018 I have an old BRAUN Plak Control Type 4726. Open it but the battery has no marks It’s an assembly of 2 cells this assembly is 56mm x 22mm Voltage mesurement 0.6v seems battery out of order. Trying to charge it with an external charger, Max voltage 2.83V. I’m waiting for 1 hour. Where i can find this kind of batteries, i’m located in France. Reply Ionic Admin 13 March 2018 Bonjour Fred, yes the Type 4726 is very old. We sell a

battery pack made of 2 cells for the Sonic Complete and some Plak Control models here. This should fit your toothbrush if you add a little padding around the battery. It is the correct voltage. Recharge it for 24 hours maximum. Reply Chris 25 June 2019 Hey Fred, Bonjour as well. How did you get the Type 4726 apart. Inquisiting minds would like to know. It looks like I have to break a tab at the back to get it apart. Thanks, Chris Reply John 3 April 2018 I have an Oral B Pulsonic not the SLIM. Is there a replacement battery available. The original liion battery has 4,1 V and 530 mAh. Its dimensions are 43mm x 18mm. Reply Ionic Admin 3 April 2018 The original Pulsonic brushes Type 3722 took a 43mm long, 14mm diameter Liion battery so you might find that your battery is 14mm diameter. Liion batteries used to be rated at their maximum voltage which is 4.1V or 4.2V.

Nowadays they are rated using their average voltage which is 3.6V or 3.7V. Either of these is suitable for replacing your Pulsonic battery. This battery on eBay should be the correct size for your Pulsonic toothbrush. It takes two people to remove the bottom to replace the batteries. Not user friendly. The tooth brush was getting sluggish, so now it doesn't work. Replaced with new batteries in the correct way as noted on the inside of bottom. Any suggestions what to do. Thanks, Barbara Reply Ionic Admin 2 July 2018 Is there any corrosion on the metal battery contacts inside the brush. These brushes take 2 AA batteries. Make sure you have used either alkaline disposable batteries or fully charged NiMH rechargeables. Also see our toothbrush care guide for tips on looking after your electric toothbrush. Reply DAV 10 September 2019 If it makes you feel any better, I feel like I am going to break the thing and get an excellent workout when I replace the batteries. I have to push down with all of my might against a solid surface to get the case back together. Getting it apart is nearly as difficult. And there is no corrosion. Reply Isabel Johnson 9 July 2018 I have a Oral B Pulsonic. I removed the screw and the cap as described below but I cannot slide the inside out after releasing the catches. What do I do wrong Please help. "The Pulsonic models have a screw holding the base cap in place, and once the cap's removed you can release the catches and slide the insides out. Be aware that the original Pulsonic models had Liion batteries rather than the lower voltage NiMH batteries used in the slim version." Reply Ionic Admin 9 July 2018 Hi Isabel, the trick is to make sure that both catches are released at the same time. Then you should be able to push the plastic headconnector up inside the casing of the handle to slide the insides out. Reply jan 17 July 2018 I just got a new Pulsonic Slim in a private sale, but box still sealed.